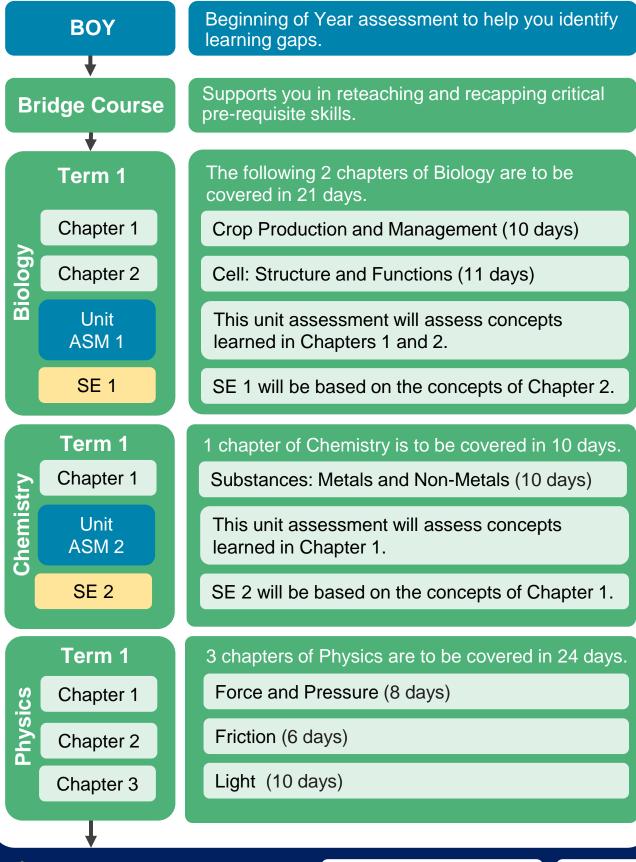
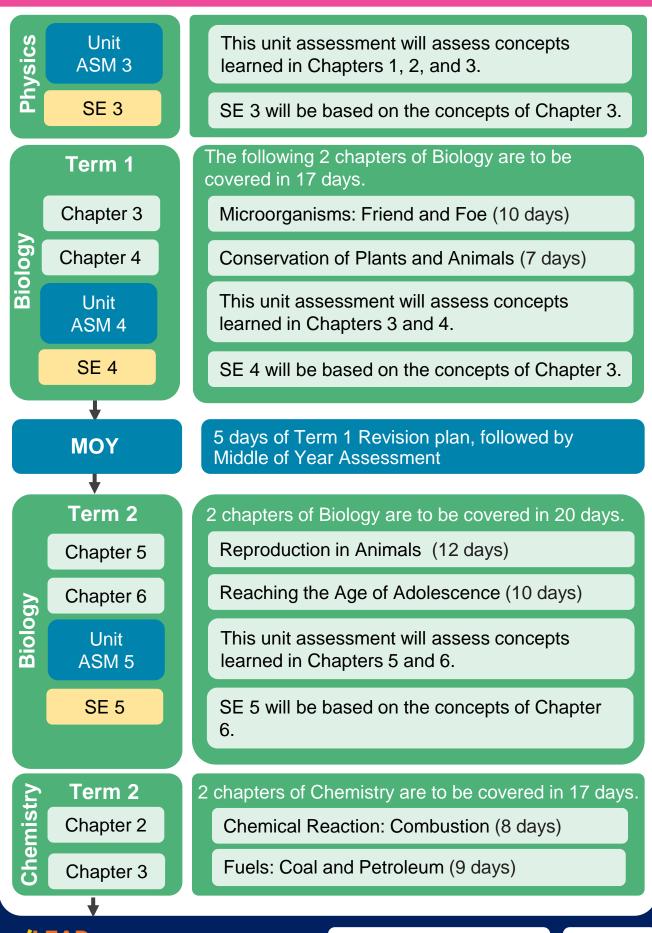
Dear teachers, the table below summarises the learning journey you will cover with your students this year.

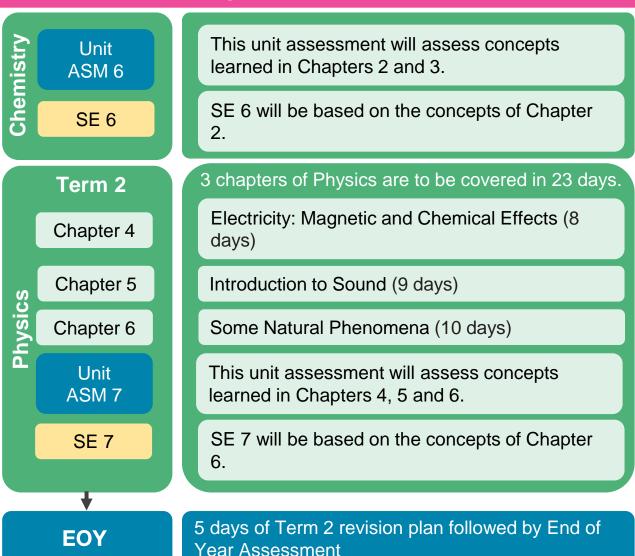




Learning Journey for the Year



Learning Journey for the Year



Note: All subject enrichment (SE) activities are optional. However, It is recommended that students perform them in class in order to strengthen their conceptual understanding.

Life Skills - The important skills that students will develop this year are:

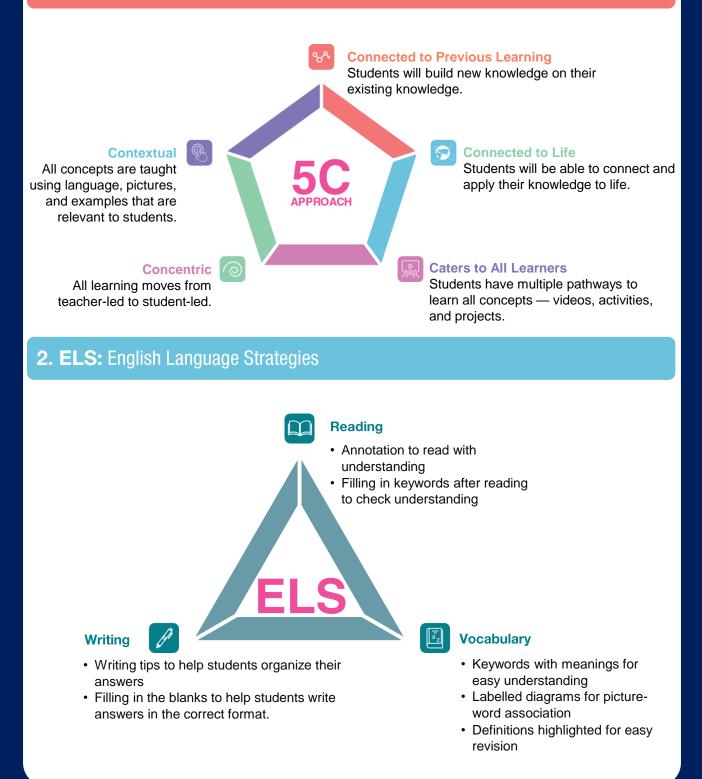
≸ THINK		ి COLLABORATE
 Solving real-world problems Creating new ideas Being curious Reflecting on your learning Learning from mistakes 	 Communicating effectively Presenting ideas Using information Using different media 	 Working with others Appreciating others' ideas Resolving conflicts Connecting yourself to your community Connecting yourself to the nation

Class 8 – Science

The LEAD Method

The LEAD Method includes unique pedagogical approaches you will use to help your students develop a deep understanding of concepts. These are integrated into the lesson plans.

1.5Cs Approach: Every concept is taught through the 5Cs approach





The LEAD Method

LBD — Learning by Doing



Learning by Doing is followed in various ways:

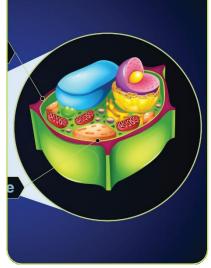
In-Class Activities



Laboratory Experiments



Videos and Demonstrations



Making Models







Important Icons and Features

Icons and Features of the Books

Provide activities and questions that help students apply new concepts to their life.

ACTIVITY

Help students understand concepts and apply their learnings.

KEYWORDS

Provide meanings of difficult words as they read.

Provide opportunities for building thinking skills.

Provide opportunities for building collaboration skills.

✤ COMMUNICATE

Provide opportunities for building communication skills.



Students can access important resources at home by scanning these codes using the LEAD Student App.

Icons and Features in the Lesson Plans



Think



Observe



Read



Turn and Talk

Turn-Write-Pair-Share

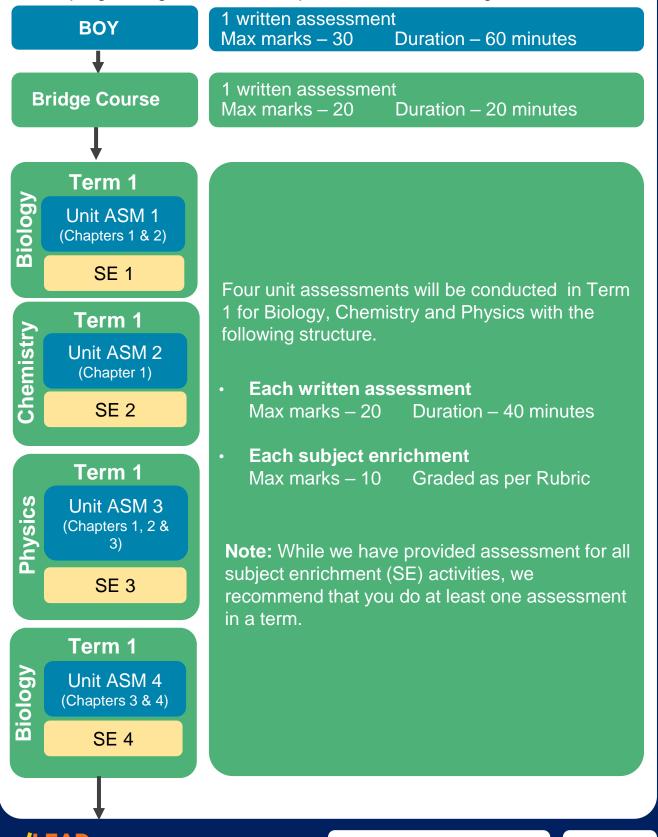
Ensure that you use the routines and structures as mentioned in the plans to achieve excellence in each unit.

Resources called LCRs will help you understand these in detail. The LCR for each routine or structure will be mentioned under 'Preparation Needed' the first few times that routine is used.



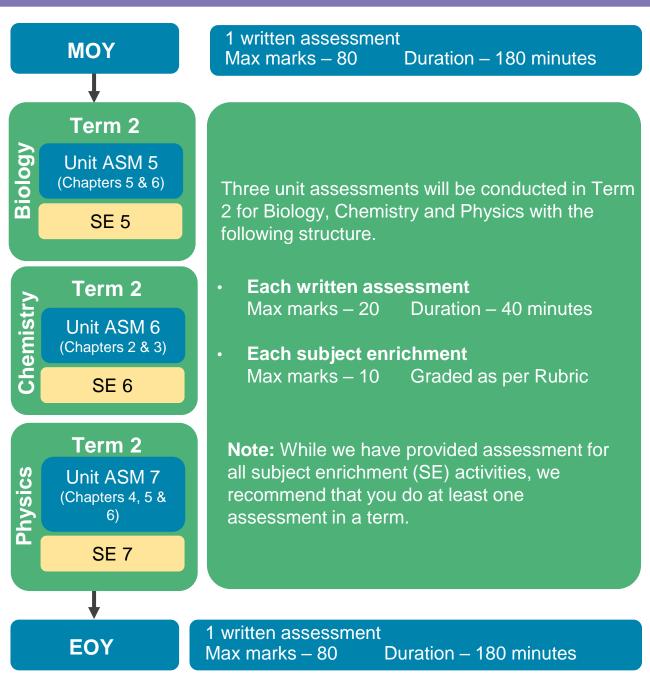
Assessment Structure for the Year

The objective of assessments is to check if all students have understood the concept and can apply their learning. Based on assessment data, it is very important to do strong remedials using LEAD remedial recommendation before progressing forward. LEAD prescribes the following assessments:





Assessment Structure for the Year





Assessment Framework

Unit Assessments

The written unit assessments have the following structure.

Types of Question	Marks	Questions	Total Marks
Multiple Choice Questions	1	4	4
Fill in the Blanks	4	1	4
Short Answer Questions	2	4	8
Long Answer Questions	4	1	4
		10 questions	20 marks

MOY & EOY Assessments

MOY and EOY assessments will have the following structure.

Types of Question	Marks	Questions	Total Marks
Multiple Choice Questions	1	16	16
Fill in the Blanks	8	2	8
Short Answer Questions	2	7	14
Short Answer Questions	3	6	18
Long Answer Questions	4	6	24
		36 questions	80 marks



Assessment Framework

Spiraling in Assessments

- In MOY 100% questions will be from Term 1 Units.
- In EOY 75% questions will be from Term 2 Units, and 25% will be from Term 1 Units.
- In Unit Assessments For every group subject, the unit assessment will cover 85%-90% marks from that unit and 10-15% marks from the previous units. This is to help students practice concepts and be better prepared for MOY and EOY.
- The exact syllabus is provided in the Important Notes of the respective assessment day.

Difficulty level of Questions

Difficulty level of questions in the assessments are based on Board guidelines. All questions are categorised as per the table below:

	LOTS (Lower Order Thinking Skills)	MOTS (Middle Order Thinking Skills)	HOTS (Higher Order Thinking Skills)
Definition	Questions based on recalling knowledge	Questions based on applying skills in familiar scenarios	Questions based on applying skills in unfamiliar scenarios, analyzing situations and building on top of what was taught in class.
Bloom's Level	Remember	Understand Application (simple)	Application (complex) Evaluate Analyse Create

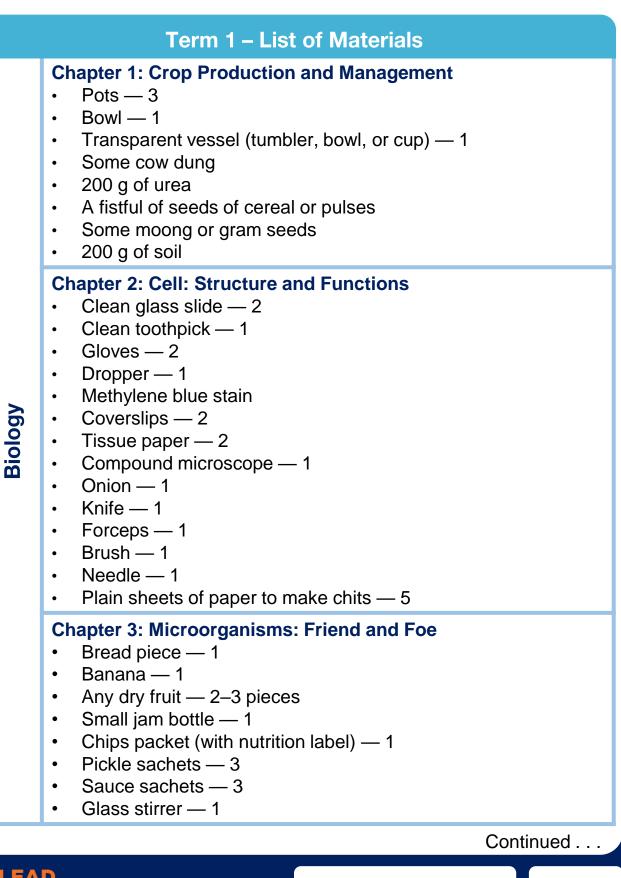
In line with Board guidelines, LEAD assessments follow the structure explained below

Unit ASM 1 - 50% LOTS : 40% MOTS : 10% HOTS Unit ASM 2 - 50% LOTS : 40% MOTS : 10% HOTS Unit ASM 3 - 40% LOTS : 50% MOTS : 10% HOTS Unit ASM 4 - 30% LOTS : 50% MOTS : 20% HOTS MOY - 30% LOTS : 50% MOTS : 20% HOTS Unit ASM 5, 6, 7 - 30% LOTS : 50% MOTS : 20% HOTS EOY - 30% LOTS : 50% MOTS : 20% HOTS

We increase the level of difficulty for students slowly in Term 1.



You will need the following materials for the various activities and experiments that will be conducted in Term 1.



Class 8 – Science

ALP

You will need the following materials for the various activities and experiments that will be conducted in Term 1.

	Term 1 – List of Materials
Biology	 Chapter 3: Microorganisms: Friend and Foe (Continued) 100 mL beaker — 1 Dropper — 1 Glass slide — 1 Coverslip — 1 Carton box or a paper bag or any open container — 1
Chemistry	Chapter 1: Substances: Metals and Non-MetalsIron nail -1 Sulphur -1 pieceElectric circuit board -1 Battery cell -1 Bulb -1 Copper wire -1 pieceMagnesium ribbon -1 Sulphur powder -10 gBlue litmus paper strips -2 Red litmus paper strips -2 A spirit lamp or a Bunsen burner -1 Test tubes -4 Glass stirrer -1 Petri dish -1 Test tube holder -1 Rubber corks -2 Matchbox -1 Pair of tweezers -1 Sodium metal -1 pieceMagnesium ribbon -1 pieceMagnesium ribbon -1 pieceMagnesium ribbon -1 pieceSulphur powder -25 gBeaker -1 Water -500 mLScalpel -1 Spirit lamp -1 Test tube stand -1



Continued . . .

You will need the following materials for the various activities and experiments that will be conducted in Term 1.

	Term 1 – List of Materials
Chemistry	 Chapter 1: Substances: Metals and Non-Metals (Continued) Spatula — 1 Sulphuric acid — 100 mL Hydrochloric acid — 100 mL Sodium hydroxide solution — 100 mL Aluminium turnings — 10 g Copper turnings — 10 g Iron filings — 10 g Graphite — 10 g Zinc pieces — 10 g Magnesium sulphate solution — 50 mL Copper sulphate solution — 50 mL Zinc sulphate solution — 50 mL Iron sulphate solution — 50 mL
Physics	Chapter 1: Force and Pressure • Bar magnets -1 • 500 mL beaker -1 • 2 litre plastic bottle -1 • Compass with a needle or a divider -1 • Balloon -1 • Tube of length 25 cm and diameter 5–7.5 cm -1 • 30 cm ruler -1 • Adhesive tape -1 • Adhesive tape -1 • Roll of thread -1 • Small table -1 • Bottle of food colour -1 • Iron nails -3 Chapter 3: Light • White sheets -3 • Plane mirrors -2 • Candle -1 • Protractor -1



ALP

Continued . . .

You will need the following materials for the various activities and experiments that will be conducted in Term 2.

	Term 2 – List of Materials
Biology	 Chapter 5: Reproduction in Animals Compound microscopes — 4 Permanent slides showing budding in <i>Hydra</i> and binary fission in <i>Amoeba</i> — 2 sets each
Chemistry	 Chapter 2: Combustion and Flame Candles — 2 A glass tube — 1 A pair of tongs — 1 A glass slide — 1 10 cm long copper wire Paper cups — 4 A 500 mL glass beaker — 1 A ceramic tile — 1 A piece of wood — 1 A glass bangle — 1 An iron nail — 1 A piece of charcoal A metal or glass container — 1
Physics	 Chapter 4: Electricity: Chemical Effects A mini-incandescent screw bulb with a bulb holder — 5 Connecting wires with stripped ends — 20–30 Two-cell battery holder (for 1.5 AA cells) — 5 1.5 AA cells — 10 Snap connector — 10 9 V battery — 10 Crocodile clips connected to wires — 10 Copper electrodes — 10 A magnetic compass — 5 A one-metre-long insulated copper wire with stripped ends — 5



Continued . . .

ALP

You will need the following materials for the various activities and experiments that will be conducted in Term 2.

Term 2 – List of Materials

Chapter 4: Electricity: Chemical Effects

- Copper sulphate salt 100 g
- Copper strip (approximately 15 × 5)
- Sandpaper 1

Physics

- A brass key 1
- A 250 mL beaker 1
- Distilled water 500 mL
- Hydrochloric acid 50 mL
- Sodium hydroxide solution 50 mL

